

ORIGINAL COURSE IMPLEMENTATION DATE: REVISED COURSE IMPLEMENTATION DATE: COURSE TO BE REVIEWED (six years after UEC approval): Course outline form version: 05/18/2018 September 1993 September 2019 February 2024

OFFICIAL UNDERGRADUATE COURSE OUTLINE FORM

Note: The University reserves the right to amend course outlines as needed without notice.

Course Code and Number: PHYS 105		Number of Credits: 5 Course credit policy (105)					
Course Full Title: Heat, Waves and Optics							
Course Short Title:							
(Transcripts only display 30 characters. Departments may recommend a short title if one is needed. If left blank, one will be assigned.)							
Faculty: Faculty of Applied and Technical St	udies	Department (or program if no department): Physics			vsics		
Calendar Description:							
An introductory non-calculus physics course	covering elect	tric circuits, wa	ves, geon	netric and wave optics, ar	nd thermodynamics.		
Prerequisites (or NONE):	(One of [Pri	nciples of Math	ematics 1	12, Pre-Calculus 12, MAT	H 093, MATH 095,		
MATH 09		96, or MATH 110] and one of [Physics 11, PHYS 083, or PHYS 100]) or (one of 12, PHYS 093, PHYS 101, or PHYS 111).					
Corequisites (if applicable, or NONE):	NONE		<u> </u>				
Pre/corequisites (if applicable, or NONE):	NONE						
Antirequisite Courses (Cannot be taken for	additional cre	ədit.)	Special Topics (Double-click on boxes to select.)				
Former course code/number: PHYS 102			This course is offered with different topics:				
Cross-listed with:			No Yes (If yes, topic will be recorded when offered.)				
Dual-listed with:			Independent Study				
Equivalent course(s):			If offered as an Independent Study course, this course may				
(If offered in the previous five years, antirequisite course(s) will be			be repeated for further credit: (If yes, topic will be recorded.)				
included in the calendar description as a note that students with cred for the antirequisite course(s) cannot take this course for further cred			No 🗌 Yes, repeat(s) 🗌 Yes, no limit				
			Transfer Credit				
Typical Structure of Instructional Hours			r credit already exists: (S	ee <u>bctransferguide.ca</u> .)			
Lecture/seminar hours		75	🗌 No 🖾 Yes				
Tutorials/workshops			Submit outline for (re)articulation:				
Supervised laboratory hours		45	□ No □ Yes (If yes, fill in transfer credit form.)				
Experiential (field experience, practicum, internship, etc			Grading System				
Supervised online activities			🛛 Lette	er Grades 🛛 Credit/No	Credit		
Other contact hours:			Maximu	um enrolment (for inform	nation only): 36		
	Total hours	120	Expect	ed Frequency of Course	e Offerings:		
Labs to be scheduled independent of lecture hours: \Box No \boxtimes Yes Annually (Every semester, Fall only, annually, etc.)					0		
Department / Program Head or Director: Norm Taylor				Date approved:	September 10, 2018		
Faculty Council approval				Date approved:	November 2, 2018		
Dean/Associate VP: Dean of Science: Lucy Lee				Date approved:	November 2, 2018		
Campus-Wide Consultation (CWC)				Date of posting:	November 30, 2018		
Undergraduate Education Committee (UEC) approval				Date of meeting:	December 14, 2018		

Learning Outcomes:

Upon successful completion of this course, students will be able to:

- Analyze circuits consisting of multiple resistive elements in series and parallel.
- Use graphical and mathematical representations to describe simple harmonic motion
- Apply the superposition principle to determine the resulting wave shapes from interfering waves in systems that involve sound waves, light waves or waves on a string.
- Solve sound and light problems that involve concepts such as intensity and the Doppler shift.
- Use geometric optics to solve problems consisting of the reflection and/or refraction of light.
- Analyze systems consisting of lenses in combination including eye glasses and microscopes.
- Identify which of the three processes of heat exchange are dominant in a given physical situation; calculate rate of heat transfer.
- Solve calorimetric problems involving temperature changes and phase changes.
- Analyze situations using the laws of thermodynamics.
- Analyze devices which convert heat into work (heat engines) or use work to remove heat from a system (refrigerators and air conditioners).

Prior Learning Assessment and Recognition (PLAR)

Yes No, PLAR cannot be awarded for this course because

Typical Instructional Methods (*Guest lecturers, presentations, online instruction, field trips, etc.; may vary at department's discretion.*) Lecture, demonstration, small group practice, discussion, laboratory.

NOTE: The following sections may vary by instructor. Please see course syllabus available from the instructor.

Author (surna	me, initials)	Title (article, book	k, journal, et	c.)	Current ed.	. Publisher	Year
1. Urone & Hinrich	IS	College Physics				Open Stax/Rice University	
2. Cutnell & Johns	on	Physics (10 th Editio	Physics (10 th Edition)			Wiley	2015
3. Knight, Jones a	nd Field	College Physics: A	College Physics: A Strategic Approach (3rd Edition)		\boxtimes	Pearson	2014
5. Walker		Physics (5 th Edition)		\boxtimes	Pearson	2016	
Required Addition Fully-equipped phys	sics lab.		are, hardwar	e, tools, specialized clo	othing, etc.)		
	45%	Assignments:	15%	Field experience:	%	Portfolio:	%
Final exam:					%		

Details (if necessary):

Typical Course Content and Topics

Weeks 1 & 2	Electric Current and Direct-Current Circuits		
Week 3	Oscillations About Equilibrium		
Week 4	Waves and Sound		
Week 5	Electromagnetic Waves		
Weeks 6 & 7	Geometrical Optics		
Week 8	Optical Instruments		
Week 9	Physical Optics: Interference and Diffraction		
Week 10 & 11	Temperature and Heat		
Week 12	Phase and Phase Changes		
Week 13	The Laws of Thermodynamics		
Laboratory Experiments			

	Laboratory Experiments	
	Experiment 1	Introduction to the Lab
	Experiment 2	Ohm's Law
	Experiment 3	Resistors in Series and Parallel
	Experiment 4	Standing Waves on a Wire
	Experiment 5	Standing Waves in an Air Column
	Experiment 6	Thin Lenses
	Experiment 7	Interference and Diffraction
	Experiment 8	Grating Spectrometer
	Experiment 9	Specific Heat of a Metal and Heat of Fusion of Water
Experiment 10 Elec		Electrical Equivalent of Heat